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SP—30—2024

FACULTY OF SCIENCE AND TECHNOLOGY

B.Pharma (Second Year) (Third Semester) EXAMINATION

MARCH/APRIL, 2024

PHARMACEUTICAL ORGANIC CHEMISTRY

Paper-II (BP-301T)

(Tuesday, 14-5-2024)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All the questions are compulsory.

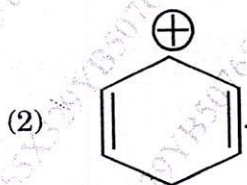
(ii) Figures to the right indicate full marks.

(iii) Draw structures wherever necessary.

1. Answer the following questions :

10×2=20

(i) Predict and justify the following structures as an aromatic/antiaromatic/non-aromatic (if any) :



(ii) Write uses of DDT.

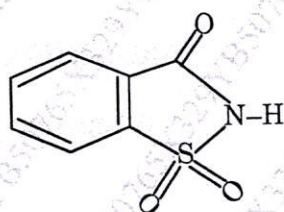
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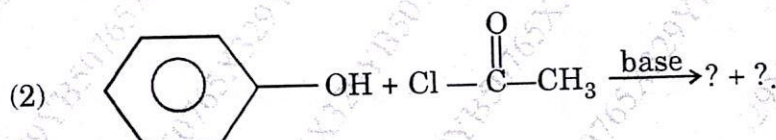
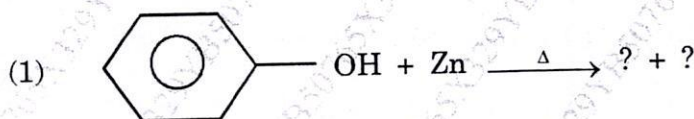
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- (iii) Draw resonance structures of anthracene.
- (iv) Differentiate between fats and oils with example.
- (v) Define Ester value.
- (vi) Write IUPAC name of the following organic compound :



- (vii) Write *one* (any) chemical reaction of aromatic amines.
- (viii) Complete the following reactions :



- (ix) Draw structures of any *two* aromatic acids and write their IUPAC names.
- (x) How to obtain phthalic anhydride from naphthalene ?

2. Answer any *two* of the following :

2×10=20

- (i) What is the effect of substituents on reactivity and orientation of monosubstituted benzene ?
- (ii) Write in detail about Bayer's strain theory.

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(iii) Write any *five* preparation methods of phenol. (Explanation is required)

3. Answer any *seven* of the following :

7×5=35

(i) Write nitration and Friedel-Crafts alkylation reactions of benzene with mechanism.

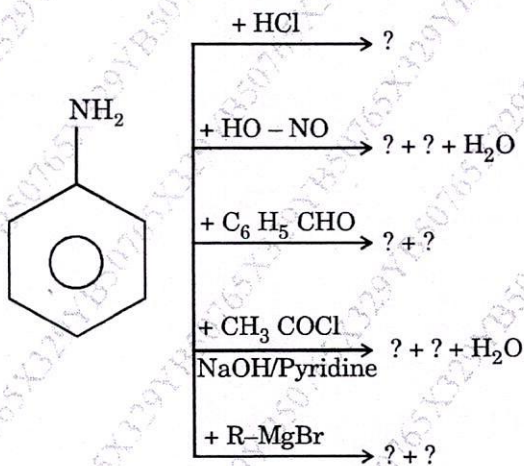
(ii) How to synthesize naphthalene by Haworth synthesis method ?

(iii) Write saponification and hydrogenation reactions of oils and fats.

(iv) Write any *two* preparation methods of cycloalkanes.

(v) Write a note on basicity of amines.

(vi) Complete the following reactions :



(vii) Write any *five* chemical reactions of benzoic acid (explanation is not required).

(viii) How E^+ attacks at C_1 , C_2 and C_3 position of anthracene.

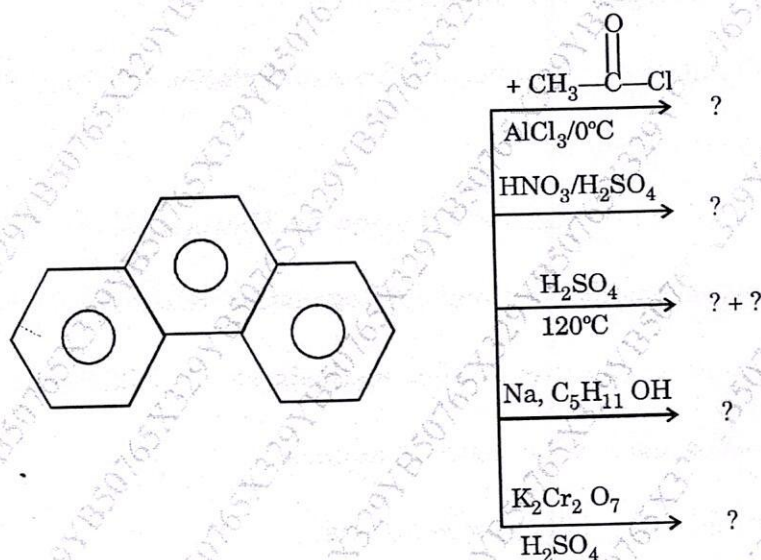
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(ix) Complete the following reactions with name of products :



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FACULTY OF PHARMACEUTICAL SCIENCES AND TECHNOLOGY

B. Pharma. (Third Semester) EXAMINATION

APRIL/MAY, 2024

PHYSICAL PHARMACEUTICS

Paper-I

(Thursday, 16-5-2024)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Figures to the right indicate full marks.

1. Answer the following :

10×2=20

- (a) Define the term surface tension and interfacial tension.
- (b) State Raoult's law. Give its significance.
- (c) Express Snell's law of refraction.
- (d) What is latent heat ?
- (e) What are buffers ? Give its applications.
- (f) Define solid, liquid and gas states of matter.
- (g) What is meant by pH and Buffer capacity ?
- (h) What is eutectic mixtures ?
- (i) Define partition coefficient. Give its applications.
- (j) Give the importance of Relative Humidity in pharmacy.

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2. Answer the following (any two) :

2×10=20

- (a) Explain various methods of determination of surface tension.
- (b) What is Solubility ? Explain various factors affecting solubility of the drug.
- (c) What is Aerosol ? Draw neat labelled diagram of aerosol. Discuss about inhalers.

3. Answer the following (any seven) :

7×5=35

- (a) Describe methods of determination of isotonicity.
- (b) What is complexation ? Give the classification and its applications.
- (c) Define HLB. Describe any *one* method to determine HLB.
- (d) What is critical solution temperature ? Explain partial miscibility curve of phenol and water system.
- (e) What is Refraction ? Describe the factors affecting Refractive index.
- (f) Explain protein binding and its applications.
- (g) What do you mean by optically active compound ? Explain its significance and limitations.
- (h) What is Sorensen's pH scale ? Describe electrometric method of pH determination.
- (i) Explain in brief about amorphous substances and polymorphism.

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FACULTY OF PHARMACEUTICAL SCIENCES AND TECHNOLOGY

B. Pharm. (Second Year) (Third Semester) EXAMINATION

APRIL/MAY, 2024

PHARMACEUTICAL MICROBIOLOGY

(Saturday, 18-05-2024)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Answer to the point only.

(iii) Figures to the right indicate full marks.

1. Answer the following questions :

10×2=20

- (a) Give the general properties of viruses.
- (b) Give the various shapes of bacteria.
- (c) Give the various functions of pili.
- (d) What are iodophors ? Give advantages.
- (e) Draw a well-labelled diagram of an aseptic area.
- (f) What are HEPA filters ?

P.T.O.

- (g) Give the clinical significance of candida albicans.
- (h) Give the contribution of Louis Pasteur in the development of Microbiology.
- (i) Define microbial spoilage of pharmaceuticals. Give examples.
- (j) Define preservatives with examples.

2. Answer any *two* of the following :

2×10=20

- (a) Explain bacterial growth curve. Write about physical conditions affecting bacterial growth.
- (b) Describe lytic and lysogenic life cycle of bacteriophages.
- (c) Explain CM test for evaluation of disinfectants. Give advantages and disadvantages of phenol coefficient tests.

3. Answer any *seven* of the following :

7×5=35

- (a) Explain the nutritional requirements of bacteria.
- (b) Explain factors affecting antimicrobial activity of disinfectants.
- (c) Explain cultivation of viruses by chick embryo method.
- (d) Classify fungi with examples.
- (e) Explain various sources of contamination in aseptic area.

- (f) Describe the following classes of disinfectants :
- (i) Acids and alkalies
 - (ii) Halogens.
- (g) Write a note on laminar air flow.
- (h) Explain direct microscopic method for bacterial growth measurement.
- (i) Explain gaseous sterilization.

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FACULTY OF SCIENCE AND TECHNOLOGY

B.Pharma (Third Semester) EXAMINATION

APRIL/MAY, 2024

PHARMACEUTICAL ENGINEERING

(BP-304T)

(Tuesday, 21-5-2024)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Figures to the right indicate full marks.

(iii) Draw well labelled diagrams wherever necessary.

10×2=20

1. Answer all questions :

- (a) What are manometers ? Enlist its types.
- (b) Draw well labelled diagram of orifice meter.
- (c) Explain any two factors affecting size reduction.
- (d) Enlist the table mentioning standards for powder as per B.P.
- (e) Enlist different applications of heat transfer.
- (f) Explain difference between drying and evaporation.
- (g) What is distillation ? State its objectives.

P.T.O.

- (h) Explain the principle of tray dryer.
- (i) Differentiate between liquid and solid mixing.
- (j) What are filter aids ? Give examples.

2. Solve any *two* of the following :

2×10=20

- (a) Explain Bernoulli's theorem. Derive its equation.
- (b) With well labelled diagram explain construction and working of liquid to liquid heat interchanger.
- (c) What is corrosion ? Explain in detail types of corrosion.

3. Solve any *seven* of the following :

7×5=35

- (a) Write a note on Reynolds number.
- (b) Write principle and construction of fluid energy mill.
- (c) Explain Fourier's law.
- (d) Explain working of simple distillation apparatus and enlist advantages and disadvantages.
- (e) Write a note on equilibrium moisture content.
- (f) With well labelled diagram explain principle and working of FBD.
- (g) Explain theories of filtration.
- (h) Write a note on perforated basket centrifuge.
- (i) Explain construction and working of rota meter with a diagram.

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FACULTY OF SCIENCE AND TECHNOLOGY

B.Pharm. (III Semester) EXAMINATION

NOVEMBER/DECEMBER, 2023

PHARMACEUTICAL ORGANIC CHEMISTRY-II

Paper BP301T

(Tuesday, 26-12-2023)

Time : 2.00 p.m. to 5.00 p.m.

Time—Three Hours

Maximum Marks—75

N.B. :— (i) Solve all questions.

(ii) Draw structure and reating wherever necessary.

(iii) Figures to the right indicate full marks.

1. Answer the following questions :

10×2=20

(a) Write structure and uses of Saccharine and BHC.

(b) Define activating and deactivating groups with examples.

(c) Write structure and uses of aryl diazonium salts.

(d) Explain synthetic uses of aryl diazonium salts.

(e) What is Huckel's rule ? Give its significance.

(f) Give the qualitative test of phenol.

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- (g) Define the terms aromaticity and resonance.
- (h) What are ortho para directing groups ? Give examples.
- (i) Discuss the stability of cycloalkanes.
- (j) Outline *two* reactions of cyclopropane.
2. Answer any *two* of the following : 2×10=20
- (a) Define electrophilic substitution reaction. Explain mechanism of nitration and sulphonation of benzene.
- (b) What are aromatics ? Explain the reactions of aromatic amines. Discuss the effect of substituents on basicity of aromatic amines.
- (c) What are fatty acids ? Explain significance and reactions of hydrolysis, hydrogenation, rancidity and drying of oils.
3. Answer any *seven* of the following : 7×5=35
- (a) Explain the Friedel-Craft's alkylation of benzene with limitations.
- (b) Outline any *two* synthesis and reactions of naphthalene.
- (c) Define angle strain. Explain why higher cycloalkanes are more stable than lower members.
- (d) Give any *four* chemical reactions of cyclobutane.
- (e) Explain the reaction and mechanism of nitration of benzene.

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- (f) Give structure and uses of :
- (i) phenol
 - (ii) *o*-cresol
 - (iii) resorcinol
 - (iv) α -naphthol
 - (v) β -naphthol.
- (g) Describe Bayer's strain theory. What are its limitations ?
- (h) Describe any *one* method to determine Reichert-Meissl value with its significance.
- (i) Explain Sacke-Mohr theory and molecular orbital concept of Cycloalkanes.

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FACULTY OF PHARMACEUTICAL SCIENCE AND TECHNOLOGY

B.Pharm. (Second Year) (Third Semester) EXAMINATION

NOVEMBER/DECEMBER, 2023

PHYSICAL PHARMACEUTICS-I

Paper BP302T

(Thursday, 28-12-2023)

Time : 2.00 p.m. to 5.00 p.m.

Time—Three Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Figures to the right indicate full marks.

1. Answer the following :

10×2=20

- (a) Define partition coefficient.
- (b) What do you mean by Glassy State ?
- (c) Define solubility.
- (d) What is meant by pH and buffer capacity ?
- (e) State Rault's law.
- (f) Give applications of buffers.
- (g) What are eutectic mixtures ?
- (h) What is latent heat ?
- (i) Define relative humidity.
- (j) What is meant by vapour pressure ?

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2. Solve any *two* of the following :

2×10=20

- (a) Discuss factors affecting solubility of drugs.
- (b) What are aerosols ? Draw a neat labelled diagram of aerosol and give its application.
- (c) Define surface tension. Explain various methods of measurement of it.

3. Solve any *seven* of the following :

7×5=35

- (a) What is refraction ? Give application of refractive index.
- (b) Describe the role of polar solvent and non-polar solvents in solubility of the drugs.
- (c) Explain partial miscibility curve of phenol water system.
- (d) Explain protein binding.
- (e) What is optical rotation ? Explain polarimeter.
- (f) Give application of complexation.
- (g) Define Isotonicity. Explain any *one* method to determine isotonicity.
- (h) Explain HLB system with neat labelled diagram.
- (i) Explain in brief solubilization and detergency.

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FACULTY OF PHARMACEUTICAL SCIENCE AND TECHNOLOGY

B. Pharm. (Second Year) (Third Semester) EXAMINATION

NOVEMBER/DECEMBER, 2023

PHARMACEUTICAL MICROBIOLOGY

Paper—(BP-303T)

(Saturday, 30-12-2023)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

- N.B. :—*
- (i) All questions are compulsory.
 - (ii) Draw neat labelled diagrams wherever necessary.
 - (iii) Figures to the right indicate full marks.

1. Answer *all* the questions :

10×2=20

- (a) Give the functions of bacterial cell wall.
- (b) Enlist the different types of culture media.
- (c) Write the principle of simple staining.
- (d) Define D-value and 2-value.
- (e) Write the importance of fungi.
- (f) Draw a neat labelled diagram of Bacteriophages.
- (g) Write ideal properties of disinfectant.
- (h) How will you validate HEPA filter ?
- (i) Enlist different chemical preservatives.
- (j) Give the principle of Autoclave.

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2. Solve any *two* :

10×2=20

- (a) Write in detail importance, scope of pharmaceutical microbiology. Enlist applied branches of microbiology.
- (b) Define sterilization. List the different methods used for sterilization with suitable example. Explain filtration sterilization.
- (c) Explain in detail multiplication of Human Viruses.

3. Solve any *seven* :

7×5=35

- (a) Explain factors affecting preservative efficacy.
- (b) Explain different applications of cell cultures in pharmaceuticals.
- (c) Explain the different sources and types of microbial contamination of pharmaceutical product.
- (d) How will you assess new antibiotics by MIC.
- (e) Differentiate between Gram-Positive and Gram-negative bacterial cell wall.
- (f) Explain air sampling methods for testing of clean and aseptic room.
- (g) Give in detail classification of disinfectants with its mechanism of action.
- (h) Describe in detail lytic growth cycle of Bacteriophage.
- (i) Give the advantages and disadvantages and applications of moist heat sterilization.

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2. Answer the following (any *two*) :

2×10=20

- (a) Give principle, construction, working, uses, advantages and disadvantages of fluidized bed dryer.
- (b) Write in detail about steam distillation method.
- (c) Write in detail about sieve shaker machine.

3. Answer the following (any *seven*) :

7×5=35

- (a) Give principle, construction, working of fractional distillation.
- (b) Write in detail about fluid energy mill.
- (c) Give factors affecting evaporation.
- (d) Give principle, construction, working of venturimeter.
- (e) Write in detail about tray dryer.
- (f) Write principle, construction and working of planetary mixer.
- (g) Explain different types of corrosion.
- (h) Give principle, construction and working of filter leaf.
- (i) Write in detail about perforated basket centrifuge.

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FACULTY OF PHARMACEUTICAL SCIENCE AND TECHNOLOGY

B.Pharmacy (Third Semester) EXAMINATION

JANUARY, 2024

PHARMACEUTICAL ENGINEERING

(BP-304T)

(Tuesday, 02-01-2024)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) Answer all the questions.

(ii) Answer to the point only.

(iii) Draw neat labelled diagram wherever necessary.

1. Answer *all* the questions :

10×2=20

- (a) What is coarse powder and very fine powder ?
- (b) What is Reynolds number ? Give its equation.
- (c) Give classification of dryers.
- (d) What is convective and diffusive mixing ?
- (e) Define terms distillation and ideal solution.
- (f) Define terms conduction and convection with *one* example.
- (g) What is filter aid ? Give its ideal properties.
- (h) Define terms filtration and clarification.
- (i) Give classification of materials used for Pharmaceutical plant construction.
- (j) Draw well labelled diagram of rotary drum filter.

P.T.O.

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FACULTY OF SCIENCE AND TECHNOLOGY

B.Pharm. (Third Semester) EXAMINATION

MARCH/APRIL, 2023

PHARMACEUTICAL ORGANIC CHEMISTRY

Paper-II (BP301-T)

(Thursday, 16-03-2023)

Time : 2.00 p.m. to 5.00 p.m.

Time— Three Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Figures to the right indicate full marks.

(iii) Draw structures wherever necessary.

1. Solve the following questions :

10×2=20

(a) Define iodine value.

(b) Write hydrogenation reaction of fats and oils.

(c) Write any *one* preparation method of phenanthrene.

(d) Draw structure of DDT and write its IUPAC name.

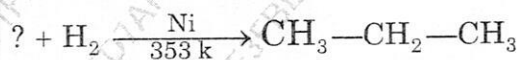
(e) Write canonical forms of anthracene.

(f) Complete the following reaction :



(g) Write uses of cresols.

(h) Complete the following reaction :



(i) Write halogenation reaction benzene.

(j) Write any *one* preparation method of cycloalkanes.

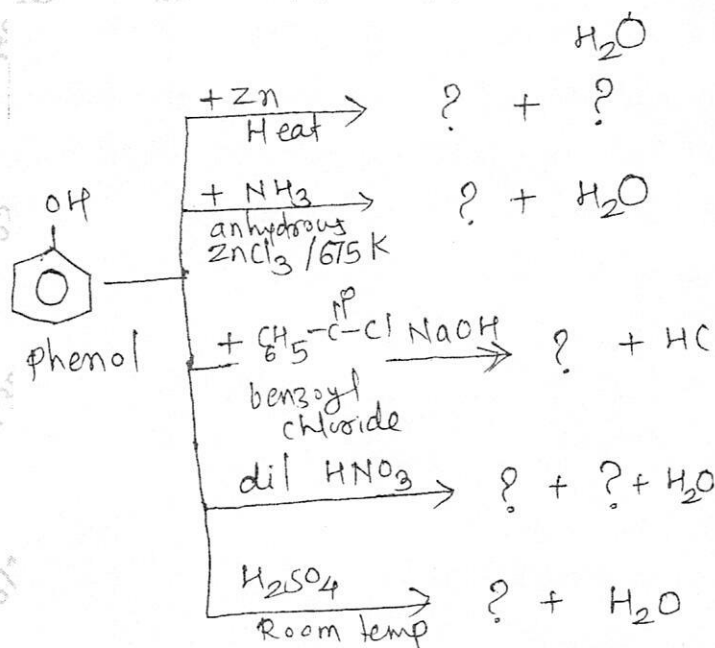
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2. Solve any *two* of the following : 2×10=20
- (a) What is the effect of substituents on reactivity and orientation of monosubstituted benzene.
 - (b) Write any *five* electrophilic substitution reactions naphthalene.
 - (c) Explain Baeyer's strain theory in detail.
3. Solve any *seven* of the following : 7×5=35
- (a) Write any *two* chemical properties of fats and oil.
 - (b) Write any *two* preparation methods of anthracene.
 - (c) Write a note on aromaticity of benzene with examples.
 - (d) Write classification and nomenclature of phenols.
 - (e) Write any *two* chemical reactions of aromatic amines.
 - (f) Why *p*-nitrobenzoic acid is stronger acid than *m*-nitrobenzoic acid ?
 - (g) Write any *two* applications of diazonium salts.
 - (h) Write notes on :
 - (a) Acid value
 - (b) Saponification value
 - (i) Complete the following reactions of phenol :



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FACULTY OF PHARMACEUTICAL SCIENCE AND TECHNOLOGY

B.Pharm. (Third Semester) EXAMINATION

MARCH/APRIL, 2023

PHYSICAL PHARMACEUTICS-I

Paper BP302T

(Saturday, 18-3-2023)

Time : 2.00 p.m. to 5.00 p.m.

Time—Three Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Figures to the right indicate full marks.

1. All questions are compulsory :

10×2=20

- (a) Define Solubility.
- (b) Define solvation and association.
- (c) Define eutectic mixture and liquid crystal.
- (d) Define surface and interfacial tension.
- (e) Define complexation.
- (f) Give any *two* applications of buffer.
- (g) What is meant by pH and buffer capacity ?
- (h) What is Raoult's law ?
- (i) What is meant by vapour pressure ?
- (j) Define spreading coefficient.

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2. Solve any *two* of the following : 2×10=20

- (a) Explain distribution law, its limitation and applications in detail.
- (b) Define Refractive Index. Explain its method of measurement, instrumentation and applications in detail.
- (c) Define surface tension. Explain various methods of measurement of it.

3. Solve any *seven* : 7×5=35

- (a) Explain C.S.T. by phenol-water system in detail.
- (b) What is meant by aerosol ? Give its applications.
- (c) Define HLB. Draw a well labelled diagrams of HLB scale and give its applications.
- (d) Explain Protein Binding.
- (e) Write about buffers in Pharmaceutical and Biological Systems.
- (f) Give applications of complexation.
- (g) Explain in brief solubilisation and detergency.
- (h) Explain in brief polymorphism.
- (i) Explain in brief determination of dissociation constant and its applications.

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FACULTY OF PHARMACEUTICAL SCIENCE AND TECHNOLOGY

B.Pharm. (Second Year) (Third Semester) EXAMINATION

MARCH/APRIL, 2023

PHARMACEUTICAL MICROBIOLOGY

Paper-BP-303-T

(Tuesday, 21-03-2023)

Time : 2.00 p.m. to 5.00 p.m.

Time— Three Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Draw neat labelled diagrams wherever necessary.

(iii) Figures to the right indicate full marks.

1. Answer all the questions :

10×2=20

(a) Define the following :

(i) Antiseptic

(ii) Sanitization.

(b) What is Mycology ?

(c) Define stain. Write the types of stain.

(d) Enlist pure culture techniques.

(e) Enlist which physical parameters required for growth of Bacteria.

(f) What is Bioburden ?

(g) Define (a) D-value (b) 2-Q10-value.

(h) Define disinfection. Write ideal properties of a disinfectant.

(i) Write about diluting fluid used for sterility testing.

(j) What is DOP test ?

2. Solve any two of the following :

2×10=20

(a) What is sterilization. Give its classifications and explain the different sterility indications with suitable example.

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- (b) Write in detail about contribution of Louis Pasteur in field of microbiology. Explain in detail growth curve of bacteria.
- (c) Enlist and explain factors affecting microbial spoilage.
- (d) Explain the different sources and types of microbial contamination of pharmaceutical products.
3. Solve any *seven* of the following : 7×5=35
- (a) Explain factors affecting preservative efficacy.
- (b) Explain in detail the classification of disinfectants.
- (c) Describe in short lytic cycle of Bacteriophage.
- (d) Explain in detail filtration sterilization method.
- (e) Differentiate between gram-positive and gram-negative Bacterial cell wall.
- (f) Explain in detail about shape and arrangement of bacteria.
- (g) How will you assess new antibiotics by MIC ?
- (h) Explain the different tests used for detection of microbial contamination in aseptic areas.
- (i) Explain sterility testing.

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FACULTY OF PHARMACEUTICAL SCIENCE

B. Pharm. (Second Year) (Third Semester) EXAMINATION

MARCH/APRIL, 2023

PHARMACEUTICAL ENGINEERING

Paper-BP-304T

(Friday, 24-03-2023)

Time : 02.00 p.m. to 05.00 p.m.

Time—Three Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Draw well labelled diagram wherever necessary.

(iii) Figures to the right indicate full marks.

1. Answer *all* the questions :

10×2=20

(i) Write *four* pharmaceutical applications of centrifugal separation.

(ii) Define coarse powder and moderately coarse powder.

(iii) Give the mechanisms of mixing in solids.

(iv) Describe in short, real solutions with its deviation from Raoult's law.

(v) Give classification of evaporators.

(vi) What is Reynold's number ? Describe its importance.

(vii) State and explain Fourier's law of heat transfer with equation.

(viii) Enlist types of glass according to Indian Pharmacopoeia.

(ix) Define Corrosion. List the types of corrosion.

(x) Enlist the factors influencing rate of filtration.

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2. Solve any *two* of the following :

2×10=20

- (i) Discuss principle, construction, working, merits, demerits and uses of Spray dryer.
- (ii) Describe metal as materials of pharmaceutical plant construction.
- (iii) Explain fluid energy mill with principle, construction, working advantages, disadvantages and applications by drawing neat labelled diagram.

3. Solve any *seven* of the following :

7×5=35

- (i) Write construction and working of cyclone separator with neat labelled diagram.
- (ii) Describe impellers used for liquid mixing.
- (iii) Explain in detail distillation under reduced pressure.
- (iv) Give principle, construction and working of evaporating pan.
- (v) Differentiate between orifice meter and venturi meter.
- (vi) Describe radiation heat transfer mechanism.
- (vii) With a neat labelled diagram describe construction and working of filter press.
- (viii) Define Centrifugation. Give advantages and disadvantages of centrifugation.
- (ix) Explain principle, construction and working of planetary mixer.

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This question paper contains 2 printed pages]

CK—02—2018

FACULTY OF SCIENCE AND TECHNOLOGY

B.Pharm. (III Semester) EXAMINATION

NOVEMBER/DECEMBER, 2018

PHARMACEUTICAL ORGANIC CHEMISTRY-II

{POC-II (BP30IT)}

(Monday, 3-12-2018)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Answer to the point only.

1. Answer all the questions : 20
 - (a) Write criteria for aromaticity with suitable example.
 - (b) Explain in brief Friedel Craft acylation.
 - (c) Give Haworth synthesis for naphthalene.
 - (d) Draw the structure and medicinal uses of DDT.
 - (e) Draw resonating structure of phenoxide ion.
 - (f) Draw the structure of chloramine and cresol.
 - (g) Define and classify fatty acid.
 - (h) Draw the structure and give the numbering to triphenylmethane and phenanthrene.
 - (i) Define Iodine value.
 - (j) Why amines are basic in nature ?
2. Solve any *two* of the following : 20
 - (a) Define and classify phenols. Explain the effects of substituents on acidity of phenol.
 - (b) Explain :
 - (i) Baeyer's Strain Theory
 - (ii) Coulson and Moffitt's modification.
 - (c) Explain the effects of substituents on reactivity of benzene. Write down any *four* electrophilic reaction of it.

P.T.O.

3. Solve any *seven* of the following : 35

- (a) Define and classify Amine. Write down the identification test for 1°, 2° and 3° amines (any *two*).
- (b) Explain the principle and significance of (any *two*) :
 - (i) Saponification value
 - (ii) RM value
 - (iii) Acetyl value.
- (c) Write the structure and uses of :
 - (i) BHC
 - (ii) Saccharin.
- (d) Discuss the qualitative test for phenol.
- (e) Explain the factors responsible for acidic character of aromatic carboxylic acids.
- (f) What is the difference between an oil and fat ?
- (g) What are polynuclear compounds ? How are they classified ? Give example of each group.
- (h) Give any *two* preparation reactions of :
 - (i) Cyclopropane
 - (ii) Anthracene.
- (i) Why benzene easily undergoes electrophilic substitution reaction than nucleophilic substitution reaction ?

This question paper contains 2 printed pages]

CK-05-2018

FACULTY OF SCIENCE AND TECHNOLOGY

B. Pharmacy (Second Year) (Third Semester) EXAMINATION

NOVEMBER/DECEMBER, 2018

PHYSICAL PHARMACEUTICS—I

(BP 302T)

(Wednesday, 5-12-2018)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Figures to the right indicate full marks.

1. All questions are compulsory.

10×2=20

- (a) What is relative humidity ?
- (b) What is isotropic and anisotropic substances ?
- (c) State Rault's law.
- (d) What is protein binding ?
- (e) Give applications of critical solution temperature.
- (f) Define critical point.
- (g) Define pH and buffer.
- (h) What is surface free energy ?
- (i) Define isotonicity.
- (j) What is refractive index ?

2. Solve any two :

2×10=20

- (a) Describe in detail about Distribution law. Give its applications and limitations.
- (b) Describe in detail about pharmaceutical and biological buffer system.
- (c) Describe in detail about methods of analysis of complexation.

P.T.O.

3. Solve any seven :

7×5=35

- (a) Describe in brief about measurement of pH by electrometric pH method.
- (b) Explain in brief Refractive Index.
- (c) Give applications of surfactants.
- (d) Describe in brief complexation applications.
- (e) What is vapour pressure ? Explain methods of measurement of it.
- (f) Give principles of diffusion in biological system.
- (g) Explain spreading coefficient in detail.
- (h) Explain factors affecting solubility of gas in liquid.
- (i) Explain dielectric constant and its applications.

This question paper contains 2 printed pages]

CK—08—2018

FACULTY OF PHARMACEUTICAL SCIENCE

B. Pharmacy (Third Semester) EXAMINATION

NOVEMBER/DECEMBER, 2018

PHARMACEUTICAL MICROBIOLOGY

(Saturday, 8-12-2018)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Draw neat labelled diagrams wherever necessary.

(iii) Figures to the right indicate full marks.

1. Answer the *all* questions :

10×2=20

- (a) Define Microbiology and describe the contribution of Joseph Lister.
- (b) Give the composition of Nutrient Agar Media.
- (c) Enlist the phase of growth curve in Bacteria.
- (d) Give the examples of Basic Stains.
- (e) Define sterilization and Disinjection.
- (f) What is dark field in compound microscope ?
- (g) Enlist the physical factors required for the growth of bacteria.
- (h) Define black fluid and white fluids with example.
- (i) What is preservative and preservation ?
- (j) Enlist the biological indicators in sterilization.

2. Solve any *two* of the following :

2×10=20

- (a) Classify the different sterilization methods with example and explain moist heat method of sterilization.
- (b) Explain historical development and scope of Microbiology.
- (c) Define Virion. Enlist the unique properties and cultivation of virus by chick embryo and animal inoculation method.

P.T.O.

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(2)

CK—08—2018

3. Solve any *seven* of the following :

7×5=35

- (a) Differentiate between prokaryotic and eukaryotic micro-organisms.
- (b) Compare Gram +ve and Gram -ve bacterial cell wall.
- (c) Explain growth curve of bacteria.
- (d) Explain microbial assay of vitamins.
- (e) Describe Sterility Test for sterile products.
- (f) What are the factors affecting bacterial spoilage of pharmaceutical products ?
- (g) Describe factors affecting disinfectants.
- (h) Write a short note on any *two* Bio-chemical tests.
- (i) Discuss the principle involved in sterilization of Dry heat and Moist heat methods.

CK—08—2018

This question paper contains 2 printed pages]

CK—11—2018

FACULTY OF PHARMACEUTICAL SCIENCES AND TECHNOLOGY

B. Pharm (Second Year) (Third Semester) EXAMINATION

NOVEMBER/DECEMBER, 2018

PHARMACEUTICAL ENGINEERING

(BP304T)

(Tuesday, 11-12-2018)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

- N.B. :—**
- (i) All questions are compulsory.
 - (ii) Figures to the right indicate full marks.
 - (iii) Draw well-labelled diagrams wherever necessary.
 - (iv) Answer to the point only.

1. Solve the following :

10×2=20

- (a) Define Reynolds' number.
- (b) Enlist mechanisms of size reduction with *one* example.
- (c) Define coarse powder and moderately coarse powder.
- (d) Define conduction and convection with *one* example.
- (e) State Raoult's law.
- (f) Define EMC.
- (g) State applications of drying.
- (h) Give the advantages of double cone blender.
- (i) Define filter aids. Give examples.
- (j) Give the classification of materials of construction.

2. Solve the following (any *two*) :

2×10=20

- (a) Describe the principle, construction, working, uses, advantages and disadvantages of ball mill.

P.T.O.

- (b) Describe the principle, construction, working, uses, advantages and disadvantages of fractional distillation.
- (c) Describe the principle, construction, working, uses, advantages and disadvantages of plate and frame filter.
3. Solve the following (any seven) : 7×5=35
- (a) Give the construction and working of horizontal tube evaporator.
- (b) Explain glass as a material of construction.
- (c) Explain the construction and working of perforated basket centrifuge.
- (d) Explain venturimeter in brief.
- (e) Give the principle and applications of planetary mixer.
- (f) Explain factors influencing evaporation.
- (g) Explain construction and working of cyclone separator.
- (h) Explain various types of impellers.
- (i) Explain construction and working of bag filter.

This question paper contains 3 printed pages]

CO—2—2019

FACULTY OF SCIENCE AND2 TECHNOLOGY

B. Parmacy (Second Year) (Third Semester) EXAMINATION

OCTOBER/NOVEMBER, 2019

PHARMACEUTICAL ORGANIC CHEMISTRY-II

(BP-301 T)

(Wednesday, 27s-11-2019)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Draw structure and reaction wherever necessary.

(iii) Figures to the right indicate full marks.

1. Answer *all* the questions :

10×2=20

(a) Why is *p*-nitrobenzoic acid stronger acid than benzoic acid ?

(b) Give the structure and uses of

(i) Saccharine

(ii) BHC

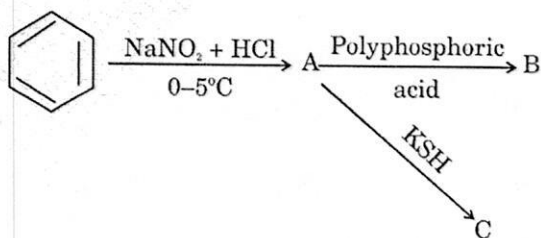
(c) Write the oxidation and reduction reaction of naphthalene.

(d) Write the difference between fats and oil.

(e) Discuss in brief Sachse-Mohr's theory.

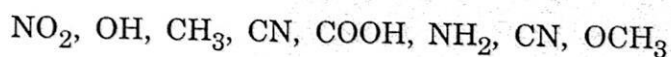
(f) Write the synthesis reaction of phenolphthalein from phenol.

(g) Complete the following reaction :



P.T.O.

- (h) Classify polynuclear aromatic hydrocarbon with suitable example.
- (i) Write the principle and significance of saponification value.
- (j) Arrange the activating and deactivating agent from the following functional group :



2. Answer any *two* of the following :

2×10=20

- (a) Define deactivating agent. Write the effect of deactivating agent on reactivity and orientation of benzene.
- (b) Write short notes on :
- (i) Kolbe's reaction and Reimer-Tiemann reaction of phenol
- (ii) Haworth reaction for synthesis of anthracene.
- (c) (i) Discuss the Baeyer's strain theory with its limitations
- (ii) Write the effect of substituent on Basicity of aromatic amines.

3. Answer any *seven* of the following :

7×5=35

- (a) Write the Friedel-Craft alkylation reaction of benzene with mechanism.
- (b) Why α -product in naphthalene is predominant over β -substitution product ?
- (c) Explain the reactivity and stability order in cyclopropane and cyclobutane with its chemical reaction.
- (d) Write the principle and significance of Iodine value and acid value.
- (e) Discuss the effect of substituent on acidity of phenol.

- (f) Write any *two* methods for synthesis of phenanthrene.
- (g) Write any *two* electrophilic substitution reactions of anthracene and mention its uses.
- (h) Write the structure and uses of the following :
- (i) Triphenylmethane
 - (ii) Resorcinol
 - (iii) DDT
 - (iv) Chloramine
 - (v) Cresols
- (i) Discuss the Huckel's rule with *three* examples.

This question paper contains 2 printed pages]

CO—07—2019

FACULTY OF PHARMACEUTICAL SCIENCES AND TECHNOLOGY

B. Pharm. (Third Semester) EXAMINATION

NOVEMBER/DECEMBER, 2019

PHYSICAL PHARMACEUTICS—I

(BP302T)

(Friday, 29-11-2019)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Answer to the point only.

(iii) Figures to the right indicate full marks.

1. Solve *all* the following :

10×2=20

- (a) What is critical micelle concentration (CMC) ?
- (b) What do you understand by the HLB system ?
- (c) State Snell's law of refraction.
- (d) What is dielectric constant ?
- (e) Define the term complexation.
- (f) What do you mean by ligands ?
- (g) What is relative humidity ?
- (h) State Raoult's law.
- (i) What are basic buffers ? Give its examples.
- (j) Give applications of buffers.

P.T.O.

2. Answer any *two* of the following :

2×10=20

- (a) What are aerosol ? Give advantages and disadvantages of aerosol.
- (b) What do you understand by optical rotation of compounds ? Give application of specific rotation in pharmacy.
- (c) Explain in detail various factors affecting the solubility.

3. Answer any *seven* of the following :

7×5=35

- (a) Define pH. Write a note on electrometric method for pH determination.
- (b) What are methods of measurement of complexation ? Discuss any *one* method in detail.
- (c) Define surface tension. Explain any *two* methods of determination of surface tension.
- (d) Define solubilization. With suitable example explain use of solubilization technique in pharmacy.
- (e) What do you mean by protein binding ? Give its application.
- (f) Give the usefulness of buffers in biological and pharmaceutical system.
- (g) What is distribution law ? Give its application.
- (h) Explain partial miscibility curve of phenol-water system.
- (i) What is detergency ? How does it take place ?

This question paper contains 2 printed pages]

CO—12—2019

FACULTY OF PHARMACEUTICAL SCIENCES

B.Pharm. (Third Semester) EXAMINATION

OCTOBER/NOVEMBER, 2019

PHARMACEUTICAL MICROBIOLOGY

(Monday, 2-12-2019)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Draw neat and well labelled diagrams wherever necessary.

(iii) Figures to the right indicate full marks.

1. Answer All questions :

10×2=20

- (a) Define D-value and Z-value.
- (b) Enlist any *four* biochemical tests.
- (c) Draw the ultrastructure of Bacteria.
- (d) What is DOP test ?
- (e) Give application of sterilization.
- (f) Write ideal properties of disinfectant.
- (g) What is spoilage ?
- (h) Define and classify the fungi.
- (i) Define stain and give its types.
- (j) Define Microbiology and mention their branches.

2. Long answer type (solve any *two* of the following) :

2×10=20

- (a) Explain cultivation of Bacteria.
- (b) Describe history, branches and scope of Microbiology.
- (c) Explain in detail the sterilization method with examples.

P.T.O.

3. Write short answer type (solve any *seven* of the following) : $7 \times 5 = 35$

- (a) Explain growth curve of Bacteria.
- (b) Write about sterility indicator.
- (c) Give the application of cell culture in pharmaceutical industry and research.
- (d) Write difference between Prokaryotes and Eukaryotes.
- (e) Explain factors affecting the microbial of pharmaceutical product.
- (f) Compare the Gram +ve and Gram -ve bacterial cell wall.
- (g) Explain sterility testing of Ophthalmic product.
- (h) Explain reproduction of virus.
- (i) Write a short note on morphology of virus.

This question paper contains 2 printed pages]

CO—17—2019

FACULTY OF PHARMACY

B. Pharmacy (Second Year) (Third Semester) EXAMINATION

OCTOBER/NOVEMBER, 2019

PHARMACEUTICAL ENGINEERING

(BP-304T)

(Wednesday, 4-12-2019)

Time : 2.00 p.m. to 5.00 p.m.

Time— Three Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Figures to the right indicate full marks.

(iii) Draw well-labelled diagram wherever necessary.

1. Answer *all* questions :

10×2=20

- (a) Give the formula of Poiseuille's equation.
- (b) Enlist modes of stress applied in size reduction.
- (c) What is size separation ?
- (d) Define conduction.
- (e) Write applications of evaporation.
- (f) Draw a well-labelled diagram of fractional distillation.
- (g) What is Raoult's law ?
- (h) Write a short note on Convective mixing.
- (i) Draw a labelled diagram of filter leaf.
- (j) What is ultracentrifuge ?

P.T.O.

2. Solve any *two* of the following :

2×10=20

- (a) Explain principle, construction and working of rotary drum filter with suitable diagram.
- (b) Describe in detail about principle, construction, working, advantages of Tray Dryer with diagram.
- (c) Discuss in brief about metal as a material.

3. Solve any *seven* of the following :

7×5=35

- (a) Write the factors influencing on corrosion.
- (b) Give the application of centrifugation process.
- (c) Discuss factors affecting on filtration.
- (d) Write about mechanism of mixing in solids.
- (e) Give the detailed classification of drying equipment.
- (f) Write the general equipment used for distillation.
- (g) Explain principle, construction of evaporating pan.
- (h) Define :
 - (i) Radiation
 - (ii) Gray body
 - (iii) Black body.
- (i) Draw a well-labelled diagram of :
 - (i) Rotary cutter mill
 - (ii) Hammer mill
 - (iii) Ball mill.